



Department of Nutrition and Food Engineering
Faculty of Health and Life Sciences
B.Sc. in Nutrition and Food Engineering
Midterm Examination Summer 2026

Course Code: 0711-2101
Level and Term: L-2, T-1
Time: 1 hour 30 minutes

Section: 252 All

Course Title: Food Chemistry
Course Teacher Initials: JR, MSZ, DMAU
Total Marks: 25

Maintaining the order of the questions is highly appreciated

- | | | Marks |
|---|---|-------|
| 1 | (a) Define food chemistry with two importance in food [CLO1,PLO1,C1] | 2 |
| | (b) Suppose you are the quality control officer of Amar Food. [CLO1,PLO1,C3]
As a quality control officer, identify the determinants of quality foods. | 3 |
| 2 | (a) Discuss the physical properties of water. [CLO1,PLO1,C1] | 2 |
| | (b) Calculate the water activity of milk powder at 25°C [CLO1,PLO1,C4]
temperature. The partial vapor pressure of water from the milk powder (P) is 15.47 mmHg, and the saturated vapor pressure of pure water (P _s) at 25°C is 23.8 mmHg. Draw a conclusion on your answer whether the milk powder is in the category of low, intermediate, or high-moisture foods. | 3 |
| 3 | (a) Construct the water activity and reaction rate [CLO1,PLO1,C3]
relationship graph for food stability. | 2 |
| | (b) How does hydrogenation occur in vegetable oils, and [CLO1,PLO1,C3]
how does it change the texture and shelf stability of food products? | 3 |
| 4 | (a) What is saturated and unsaturated fatty acids? [CLO1,PLO1,C2] | 2 |
| | (b) Identify the significance of the following tests in fats: [CLO1,PLO1,C3]
(i) Acid Value
(ii) Peroxide Value
(iii) Saponification Value | 3 |
| 5 | (a) Define carbohydrates with example. [CLO1,PLO1,C1] | 1 |
| | (b) What do you understand by epimer and isomer of sugar? [CLO1,PLO1,C3]
Provide one example each of an epimer and an isomer of sugar, and draw their structures. | 2 |
| | (c) Differentiate between amylose and amylopectin based on [CLO1,PLO1,C4]
their structural properties. | 2 |